

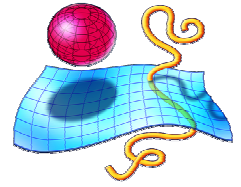
Structural Study of the Influence of Partially Crystalline Copolymers on Paraffin Crystallization in Dilute Solutions with Small Angle Neutron Scattering

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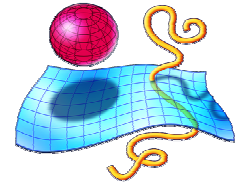
Paraffin Waxes - Properties



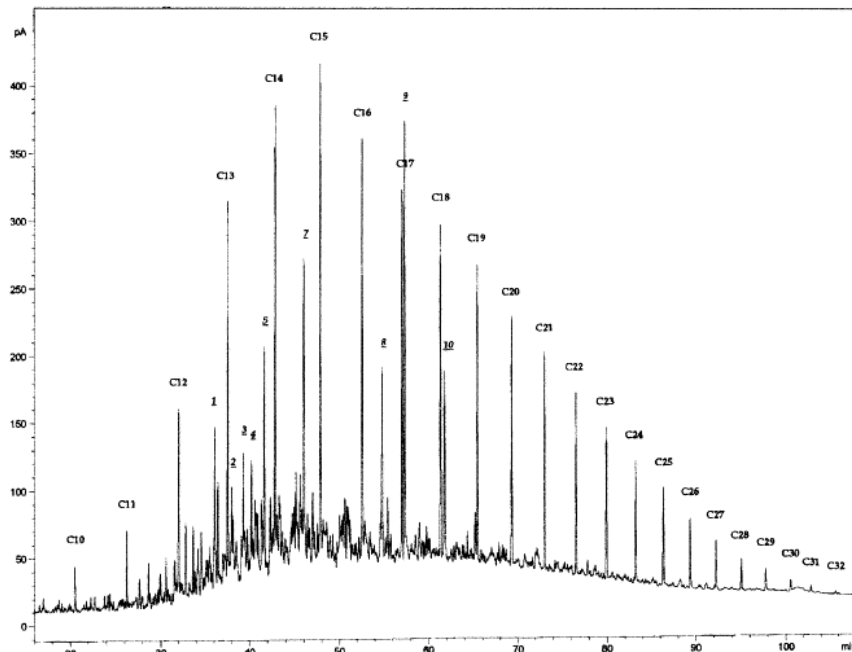
- alkanes (paraffins) : straight chain molecules, single C - C bonds: C_nH_{2n+2} or simply C_n
- gaseous: $C_1 - C_4$
- liquid: $C_5 - C_{17}$
- solid: C_{18} ($T_m \cong 28.2^\circ\text{C}$) and n larger → known as “waxes”
- paraffin waxes: orthorhombic, hexagonal, monoclinic and triclinic structures



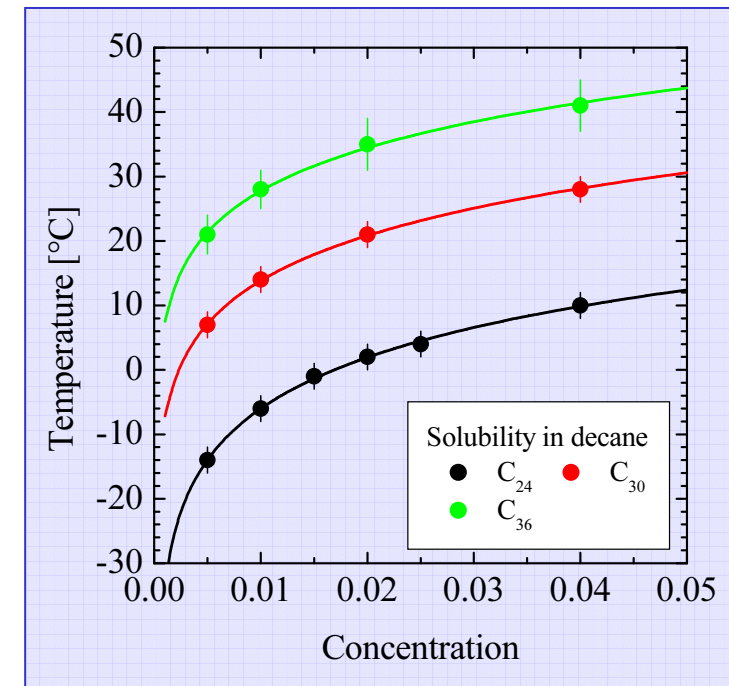
Composition of Diesel Fuel



Chromatography analysis of Diesel fuel (C_{12} and larger)



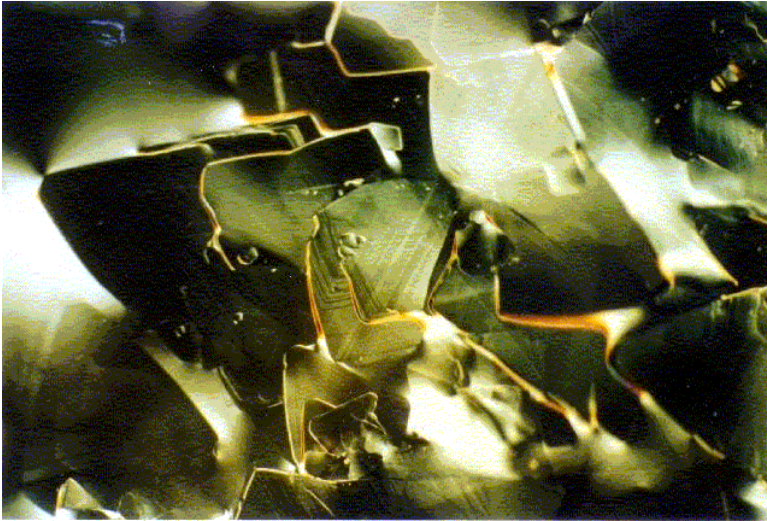
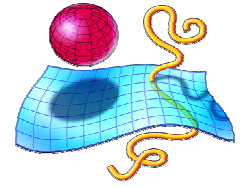
Wax solubility lines in decane (C_{10} ; $T_m \cong -30^\circ\text{C}$)



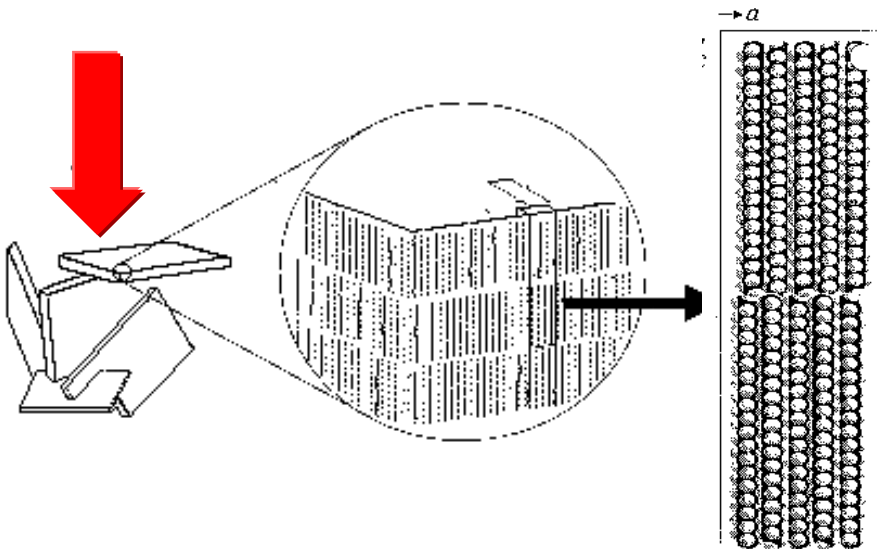
Paraffins: - important component of middle - distillate fuels (10-35%)
- major components of crude oils



Crystallization of Paraffin Waxes

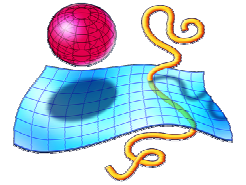


- Wax crystals form gels and stop flow
- Gels inhibit crude oil recovery from deep sea reservoirs
- Wax crystals in Diesel oil plug filters





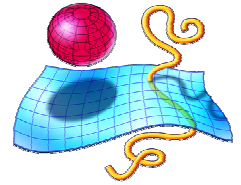
Cold Flow Properties of Middle-Distillate Fuels



- the cloud point (**CP**) - temperature when wax crystals appear (**WAT**)
- the pour point (**PP**) - temperature when fuel gels
- the cold filter plugging point (**CFPP**) - temperature when a 45 mm filter is plugged in standard condition
- typical diesel fuel without additives → **PP** \cong 10°C



Polymeric Additives



→ Crystal growth of wax

- sensitive to impurities, e.g. *modify size and shape of wax crystals*
- when intentionally added \Rightarrow impurities become additives

→ Promoters

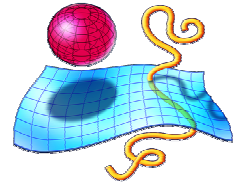
- crystallize prior to paraffins
- useful for the separation of waxes

→ Inhibitors

- wax dispersants/flow improvers (WDFI)
- crude oils and diesel fuels
 - remain fluid at low temperature and
 - pass through filters



Polymeric Additives



Choice largely by trial and error

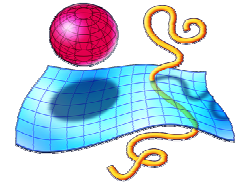
- polymers in general self assemble
- interplay between polymer aggregate and wax crystallization important

Example: poly(ethylene-co-vinylacetate) - EVA

- poorly characterized
- not very efficient in certain oils
- 50% precipitation already at high T

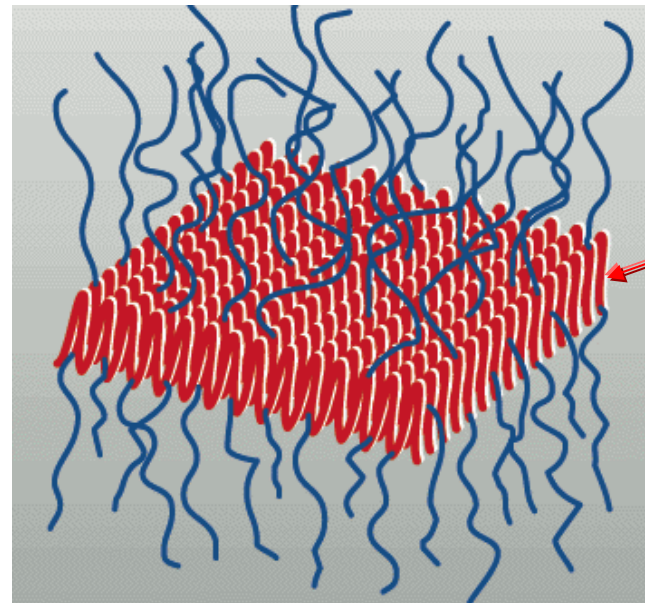
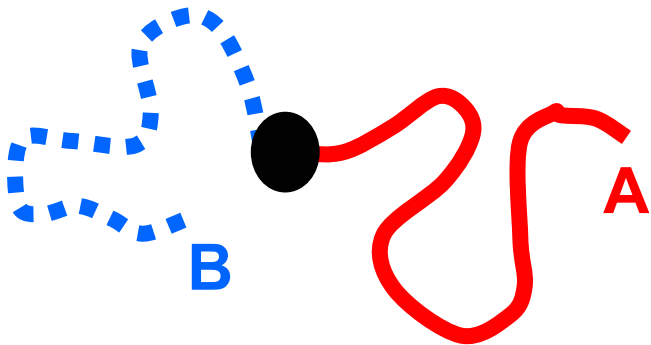


Polymeric Additives explored with SANS



1. Diblock Copolymers

PE – PEP



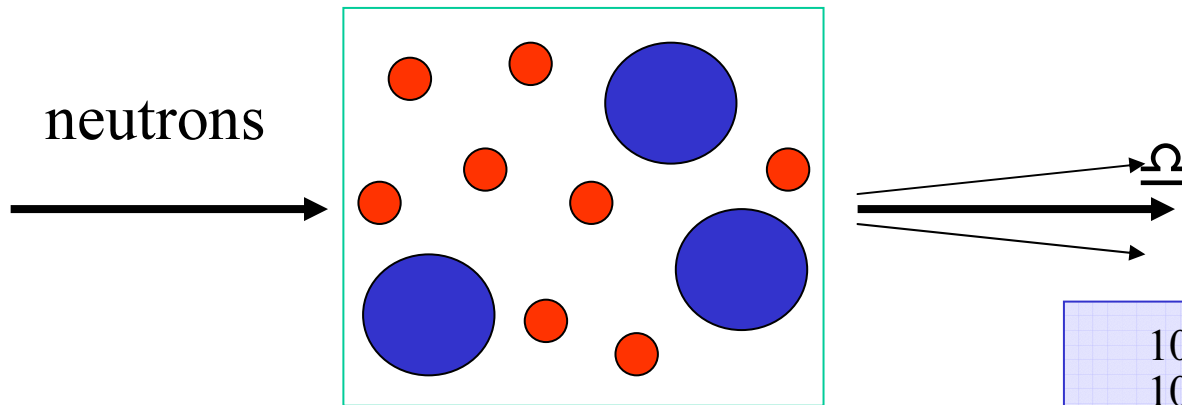
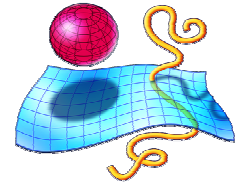
Nucleation
site

2. Random Copolymers

PEB-n type (ethylene/butylene copolymers)

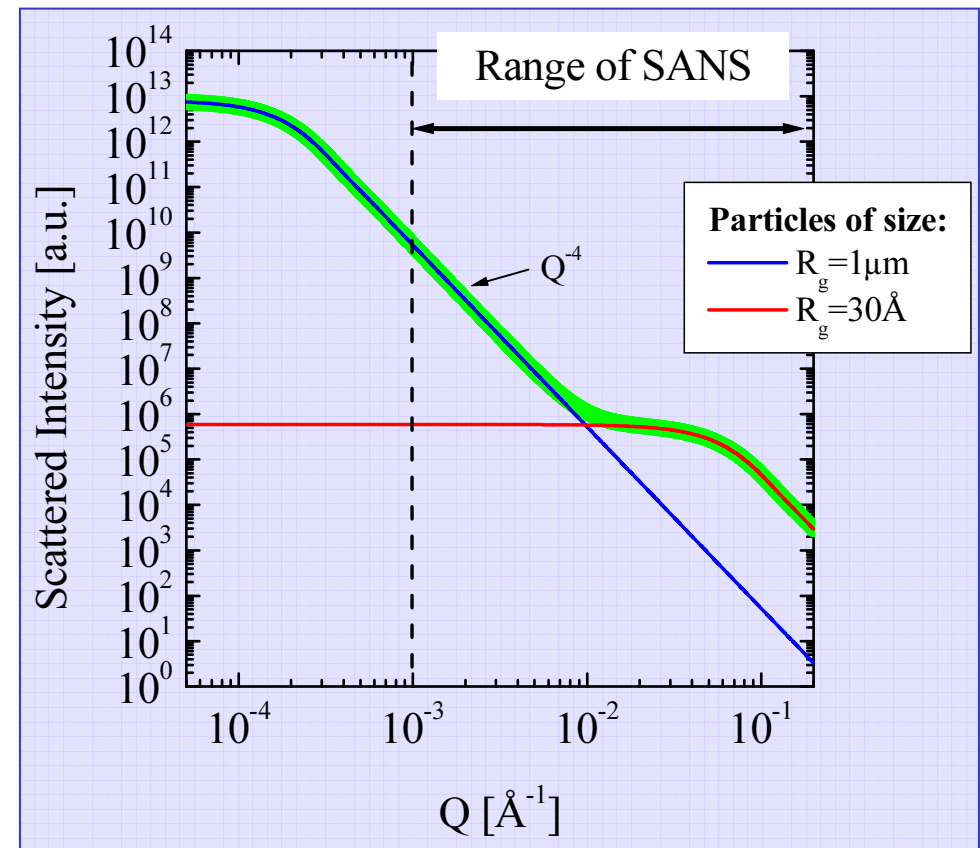


General Remarks on SANS experiments



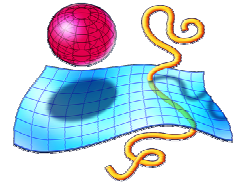
Wave number:

$$Q = (4\pi/\lambda) \sin(\theta/2)$$





Range SANS Instruments



According to $D \propto 1/Q$ one can explore particles of size:

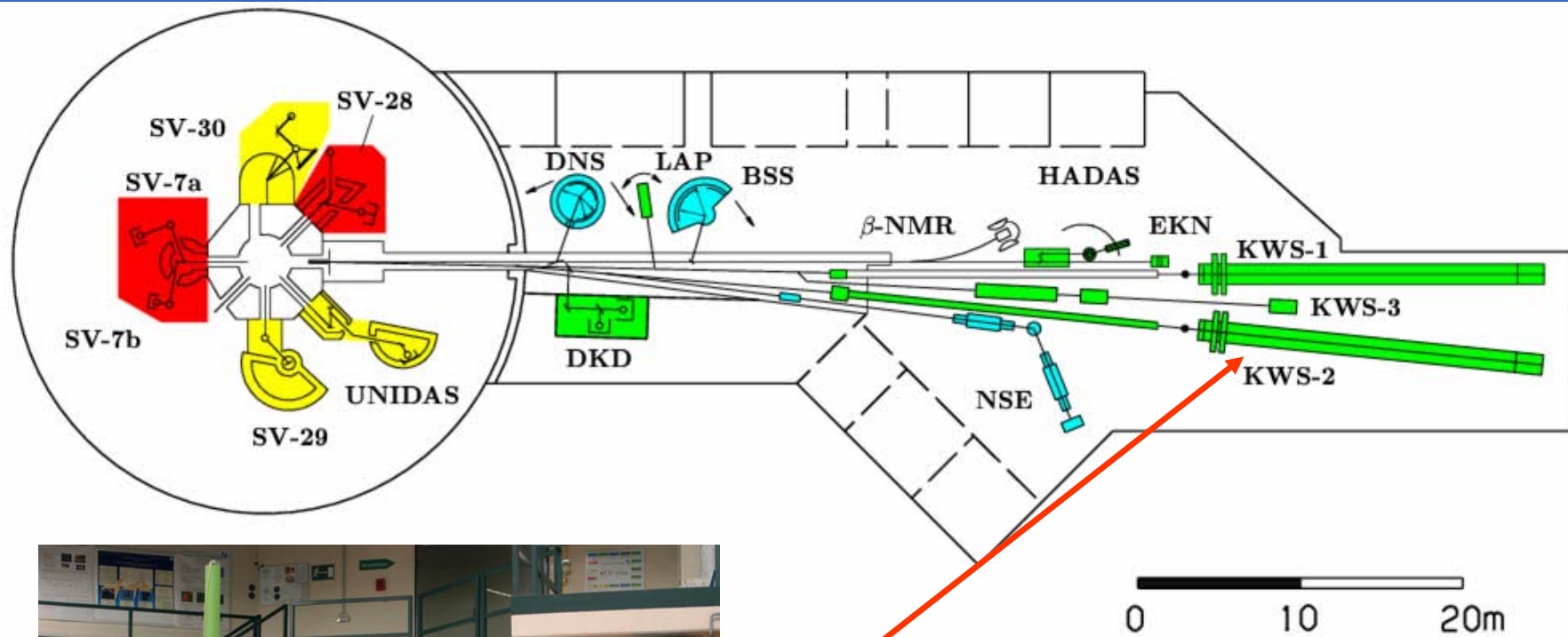
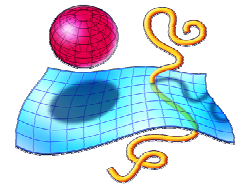
Light ($\lambda \approx 6000 \text{ \AA}$)	$0.5 \text{ }\mu\text{m} \leq D \leq 20 \text{ }\mu\text{m}$
Neutrons ($\lambda \approx (2 - 15) \text{ \AA}$)	$10 \text{ \AA} \leq D \leq 20 \text{ }\mu\text{m}$

Three different SANS techniques allow to measure a particle size in a range of four orders of magnitude ($Q \approx 2\pi/D$):

Pin - Hole SANS	$0.2 \text{ \AA}^{-1} \approx Q \approx 10^{-3} \text{ \AA}^{-1}$
Focusing SANS	$5 \cdot 10^{-3} \text{ \AA}^{-1} \approx Q \approx 10^{-4} \text{ \AA}^{-1}$
Double Crystal Diffractometer	$10^{-3} \text{ \AA}^{-1} \approx Q \approx 2 \cdot 10^{-5} \text{ \AA}^{-1}$

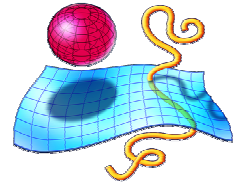


Instrumentation at FRJ-2 in Jülich

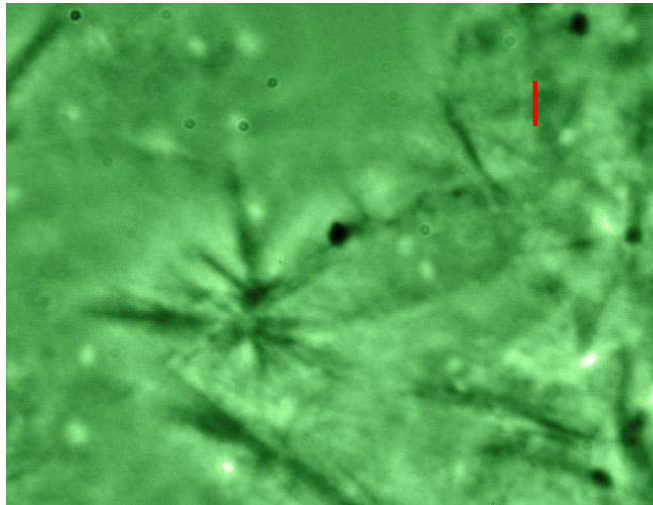




Crystallization of *syndio*-Polypropylene in d-22 solution - large scale view

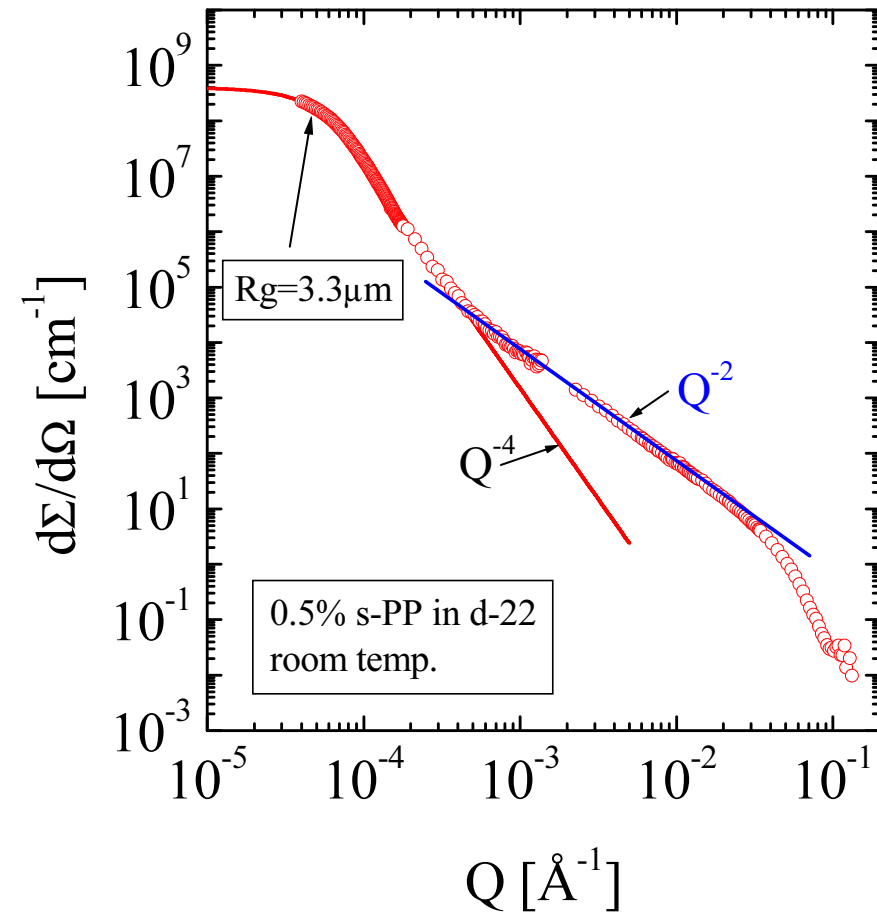


Spherulitic morphology



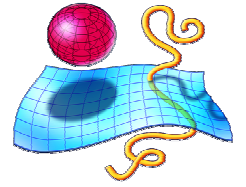
(Scale bar 2 μm)

DKD, Focusing SANS, Pin - Hole SANS

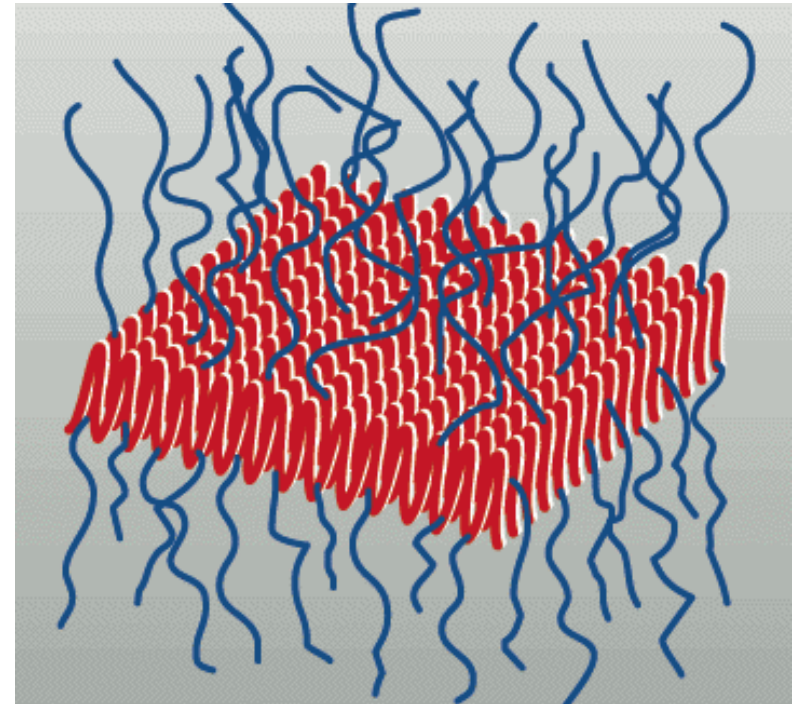
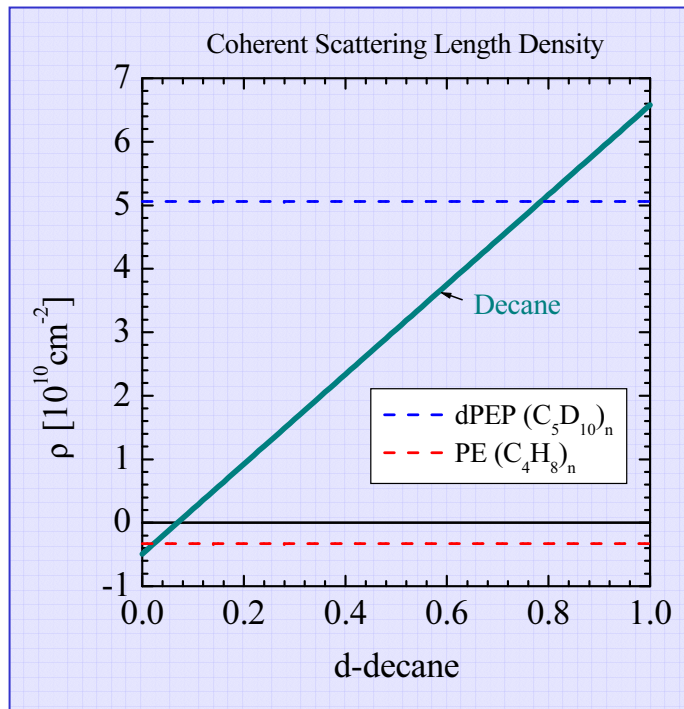




PE-PEP Diblock as Wax Crystal Modifiers



Contrast variation



Core Contrast =

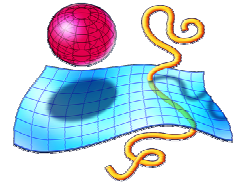


Brush Contrast =

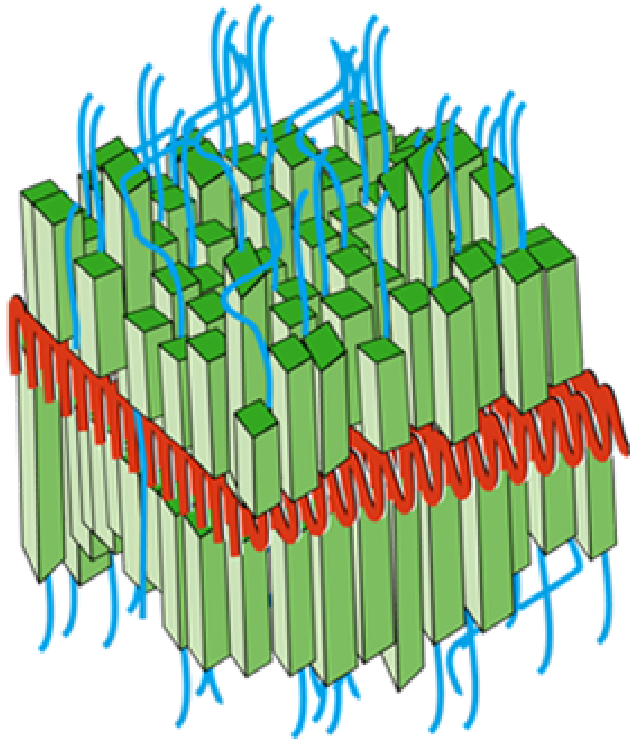




PE-PEP Diblock as Antifreeze for Diesel



polymer aggregates work
as nucleation agents



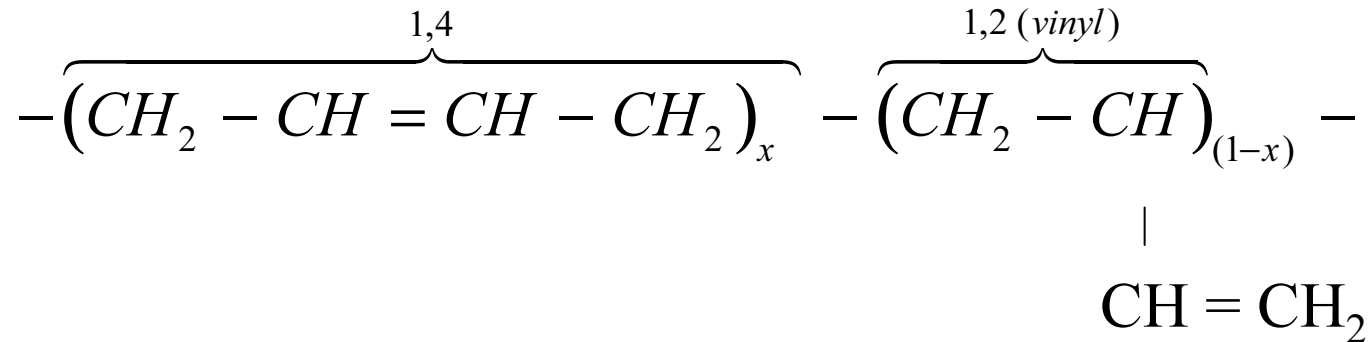
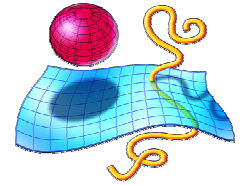
suppression of
large crystals



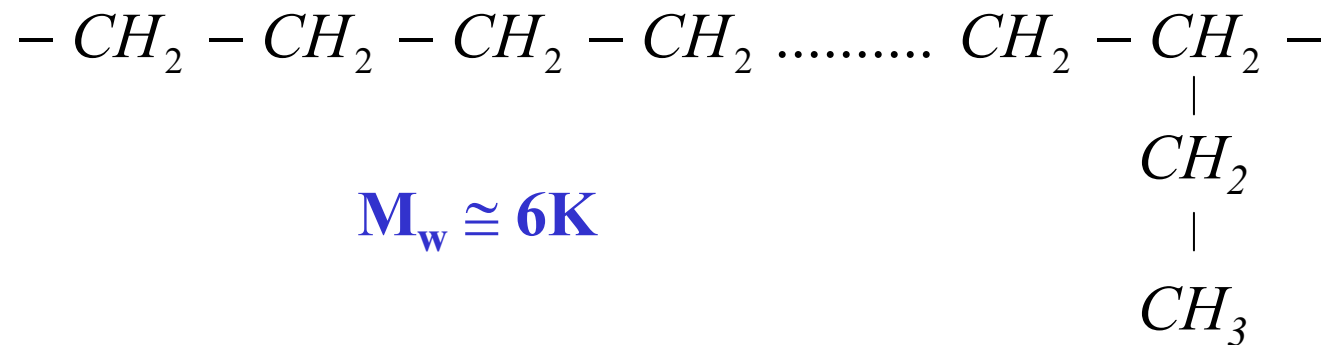
4 years from discovery to commercialization



Random Copolymers of the PEB-n Type



**PEB-n
precursor**



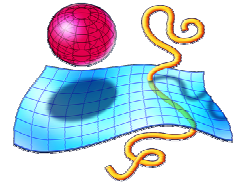
$M_w \cong 6K$

**PEB-n
n: number of
ethyl branches/ 100
backbone carbons**

**Model system for studying co-crystallization of paraffin
and polymer additives in fuel oil at low temperatures**



Scattering from Objects with characteristic Morphologies

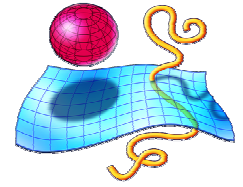


$$d\Sigma / d\Omega(Q) \propto Q^{-\alpha}$$

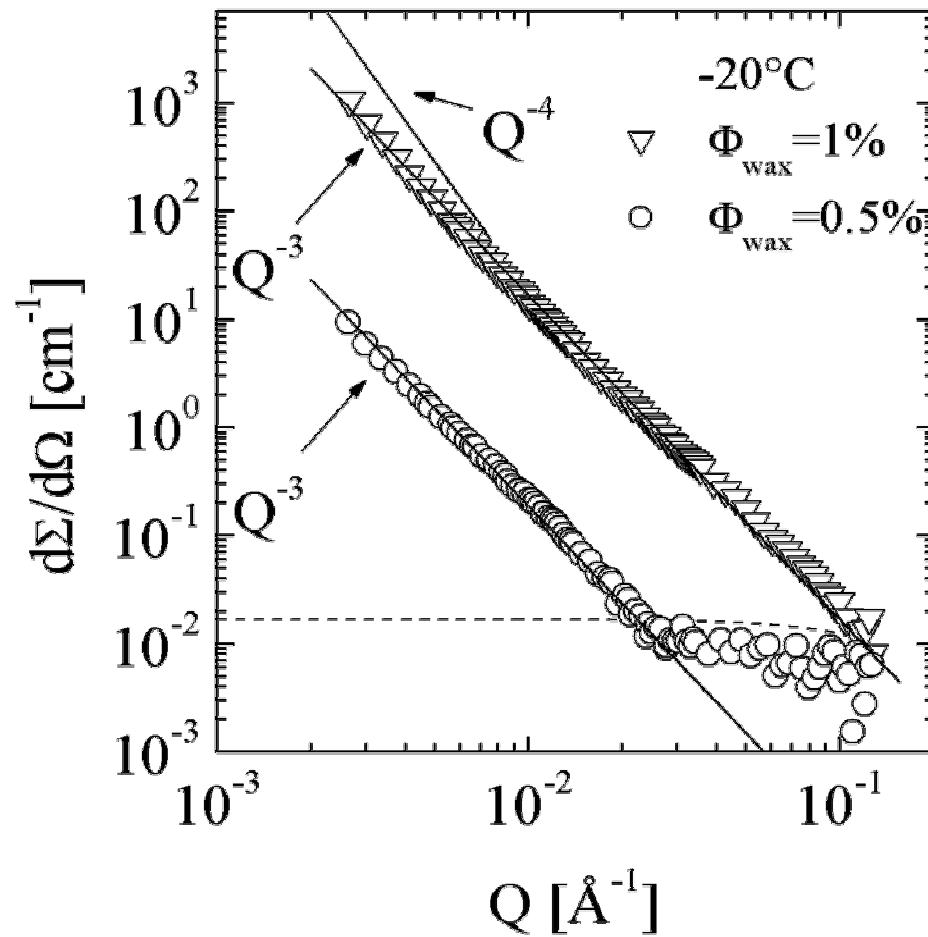
α	characteristic morphology
3/5 (Flory exponent)	swollen chain in good solvent
2	swollen chain in Θ -solvent
1	rods
2	plates
4	3-D objects ($R > 1/Q$)
< 3	mass fractals
$3 < \alpha < 4$	surface fractals
> 4	diffuse interfaces



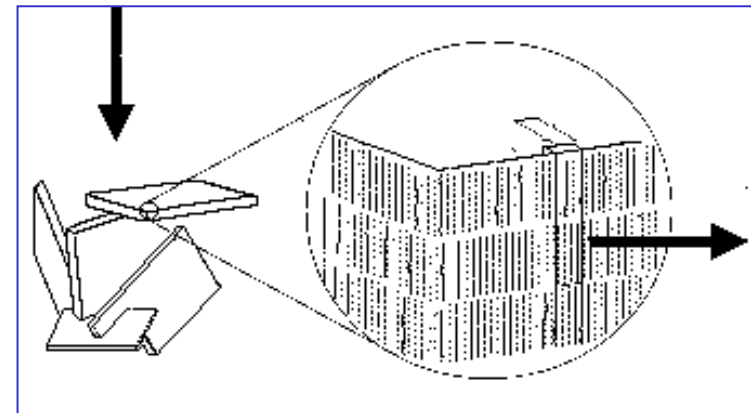
Crystallization of pure Wax



C_{24} in decane

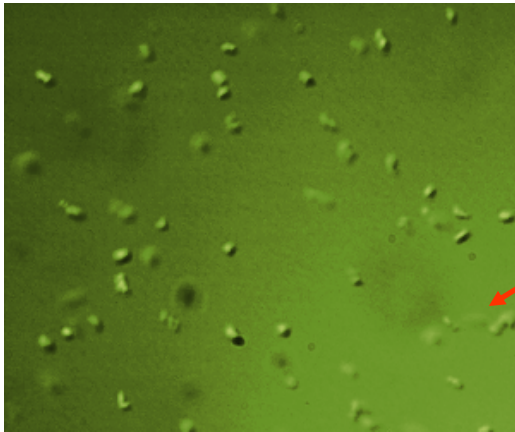
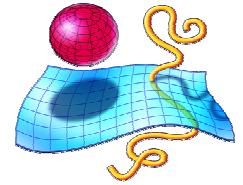


mass fractal: card house



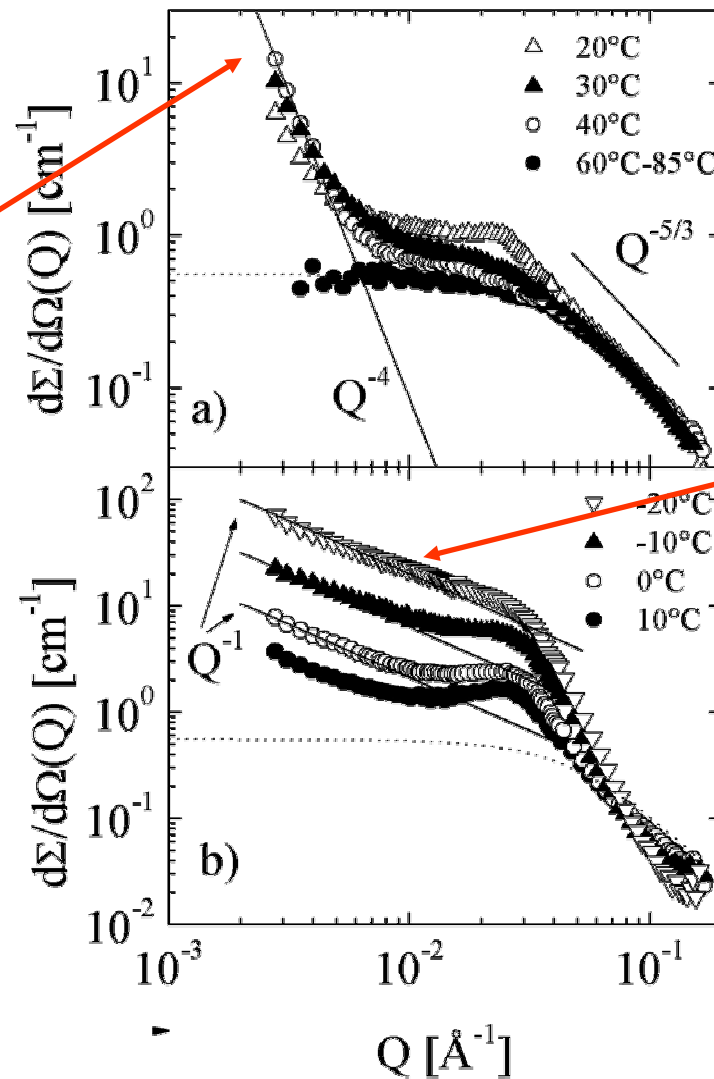


Aggregation Behavior of PEB-7.5 ($\phi_{\text{pol}}=1\%$)



Large compact objects at
by minority

From Porod constant and
micrograph 0.7% polymer
participation

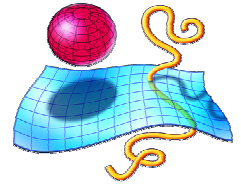


Rod like structures at
lower temperatures

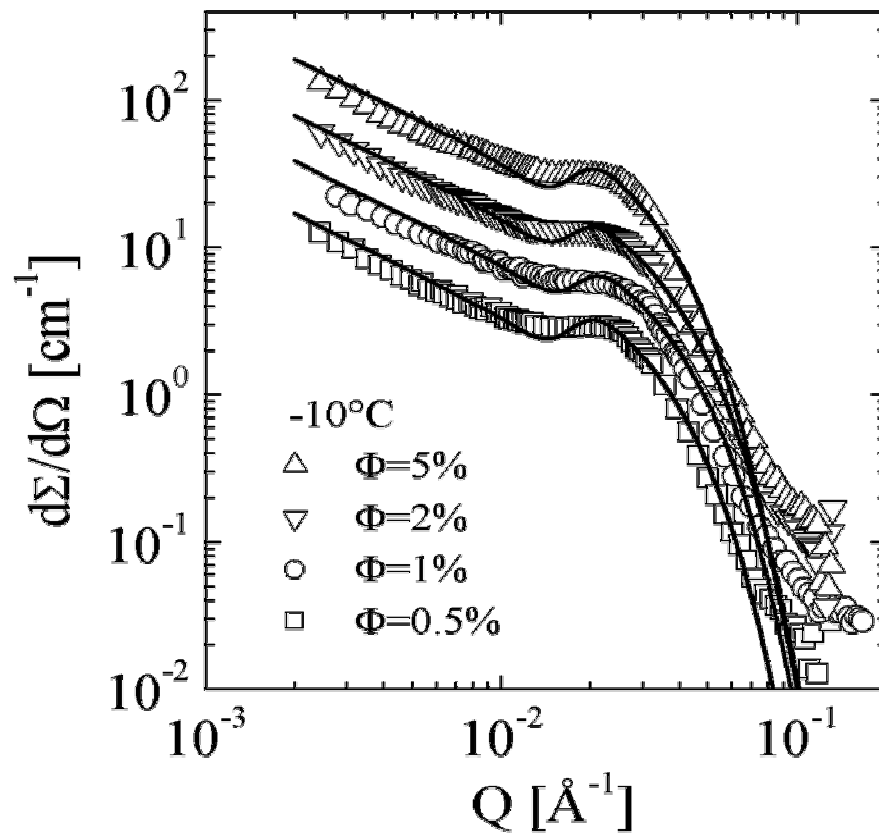
Correlation peak?



Aggregation Behavior of PEB-7.5



concentration dependence

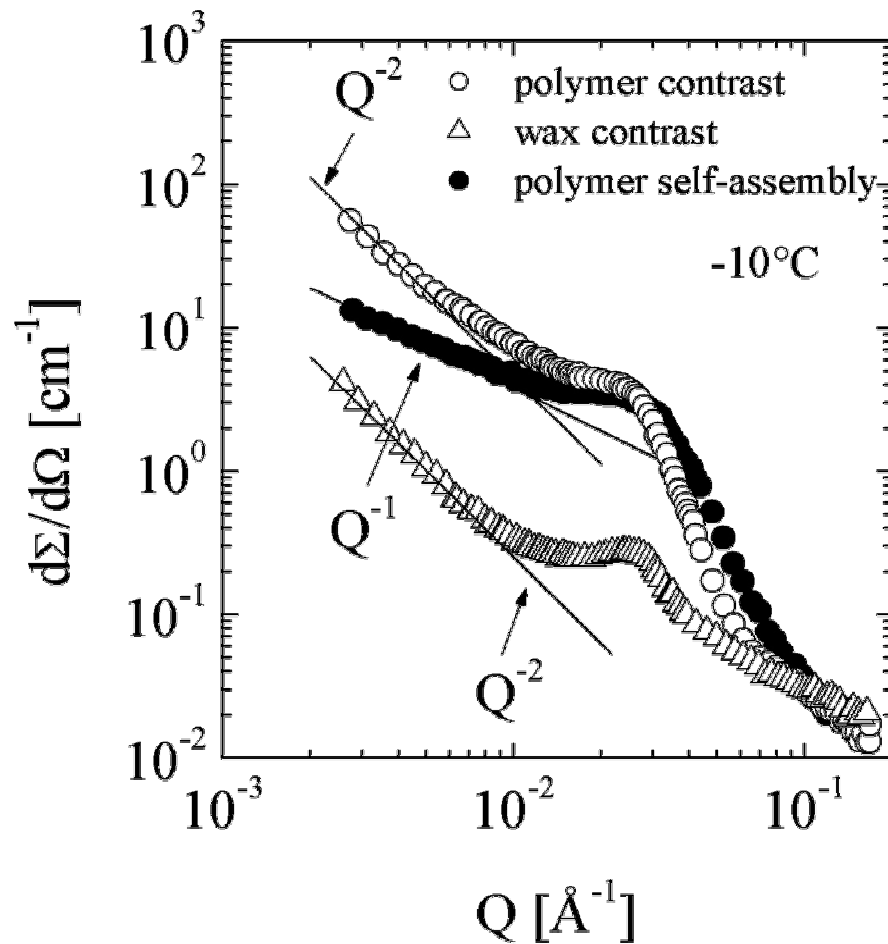
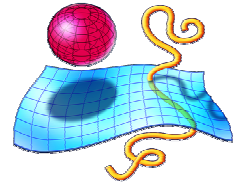


Rod like structure

Density modulation along the rod of ~ 350 Å length; Polymer density within the rod: $4\% < \phi_{\text{pol}} < 30\%$



PEB-7.5 affects C₂₄ Crystallization



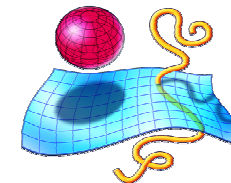
→ Aggregate morphology changes from 1D → 2D

→ Primordial modulation of the PEB 7.5 structure remains

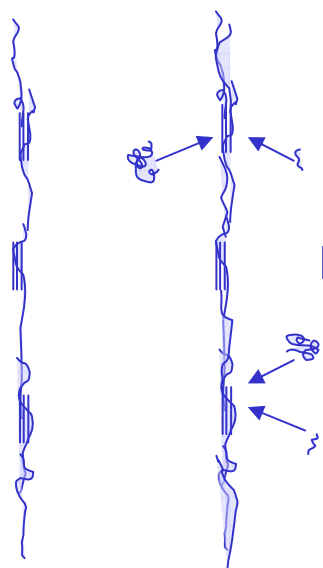
PEB 7.5 self-assembles first



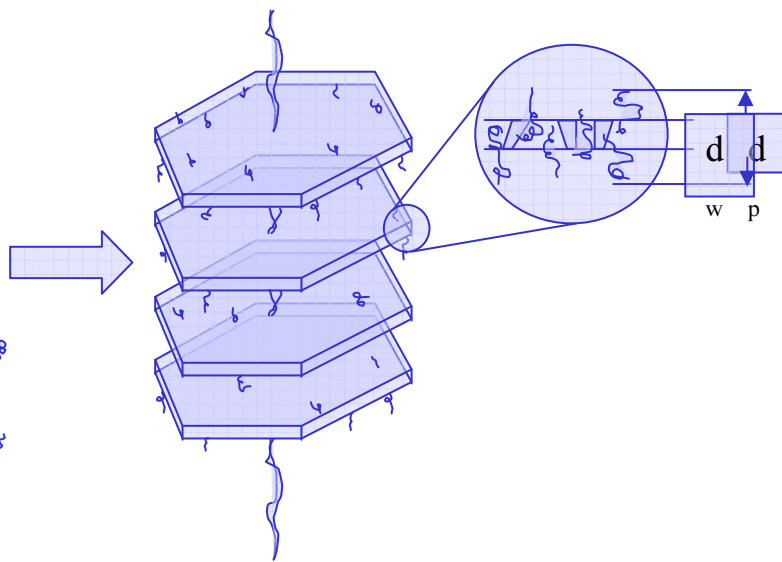
PEB-7.5 and C₂₄: Co-crystallization in correlated Plates



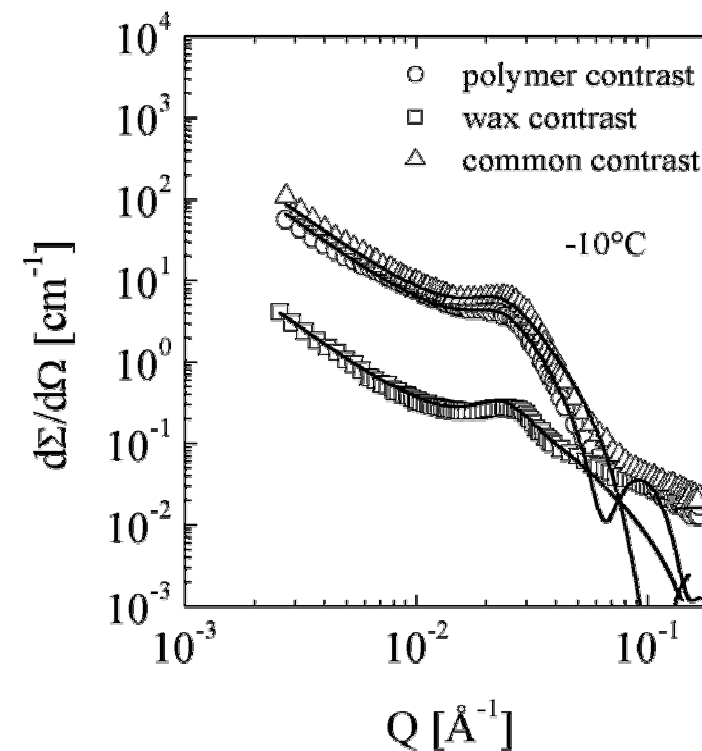
linear
morphology



correlated
plates
containing wax
+ polymer



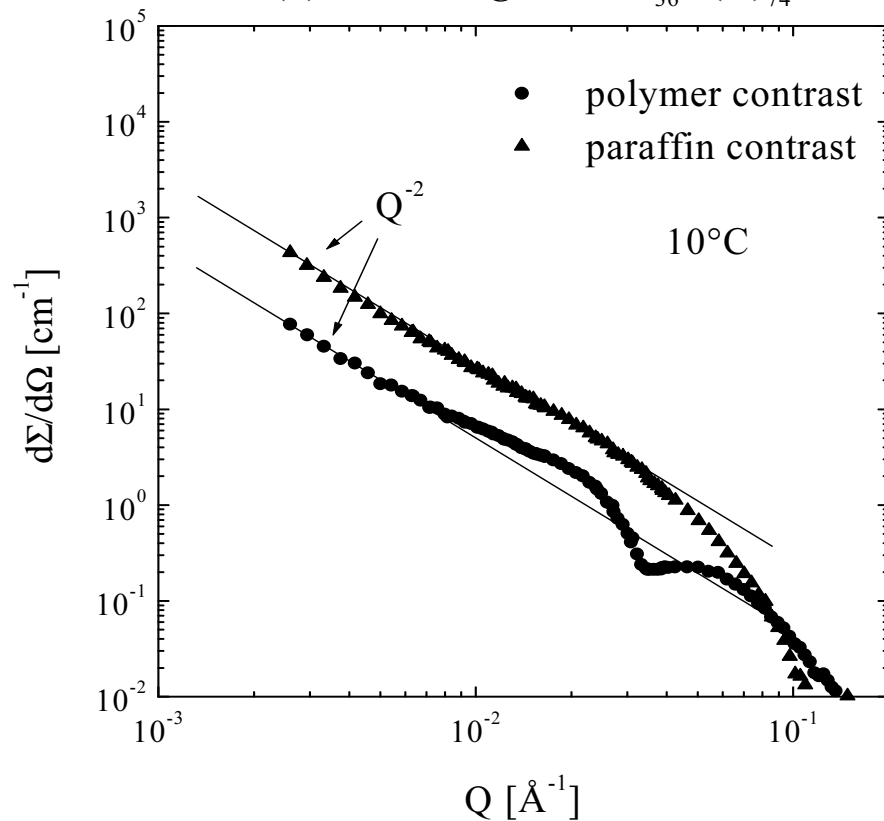
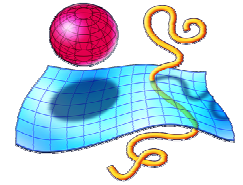
Model yields
quantitative fits



Shish-kebab morphology: similar to PE crystallization under flow



PEB-7.5 und C₃₆

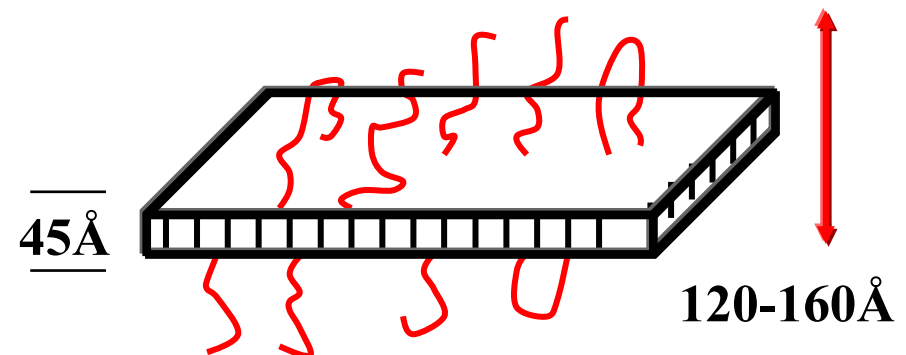


wax plates

- thickness $d_{\text{wax}} < 45\text{\AA}$
- single C₃₆ plates

polymer plates

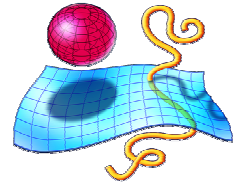
- no correlation peak
- brush like form factor



T=10°C below wax solubility line; wax co-crystallizes with polymer and suppresses the pure PEB-7.5 aggregation features



Wax Modification Efficiency



Results from evaluation of structures under different contrast

Example: wax and polymer content in platelets at 0°C, $\epsilon_w = 2\%$, and $\epsilon_{pol} = 0.6\%$

	ϵ_w wax	ϵ_{pol} pol
C ₃₆	91%	9%
C ₂₄	36%	64%

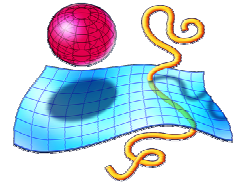
PEB-7.5 optimized for C₃₆; PEB-11 would be the optimized polymer for C₂₄

Take home message:

Efficiency highest if polymer and wax jointly crystallize

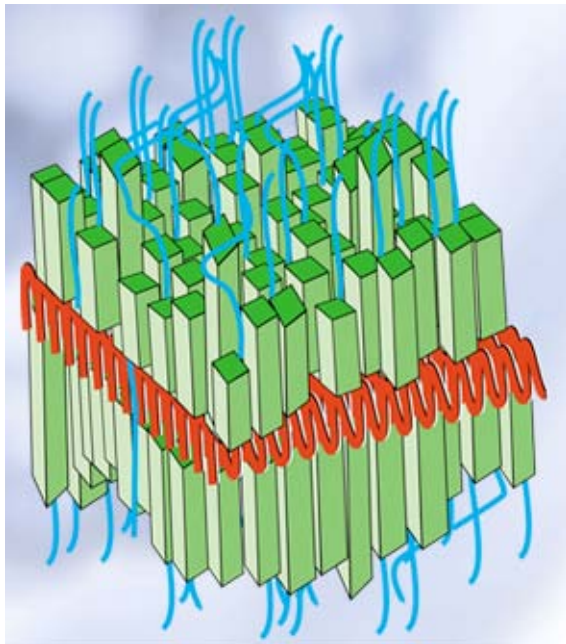


Conclusions



1. Crystalline – amorphous blockcopolymers

- polymers form templates



polymer templates nucleate wax crystals

PE-PEP is commercialized as

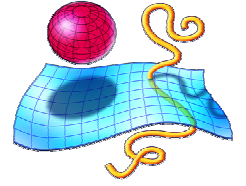
ParaflowTM

Diesel fuel additive

Wax crystal modification by partially crystalline polymers



Conclusions



2. Random crystalline amorphous copolymers

A. Polymer aggregation commences at temperatures above wax solubility line

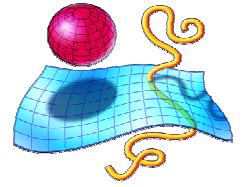
polymer form templates for crystallization

nucleation from the polymer structures

Limited efficiency



Conclusions



B. Wax solubility line in the order of polymer aggregation temperature

=> co-crystallization dominates wax

PEB-7 + C₃₆

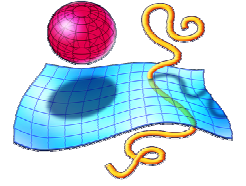
- plates at high T , suppression of 3-d objects
- polymer incorporated in wax plates

nucleation is
mediated by the
joint polymer
wax structure

High efficiency



Conclusions



3. Pour point reduction

- most efficiency if polymer and wax crystallization coincide
- tuning is possible by changing PEB-n microstructure